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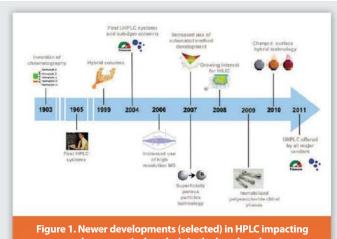
# Newer Developments in HPLC Impacting Pharmaceutical Analysis: A Brief Review

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## Introduction

High performance liquid chromatography (HPLC) is the premier analytical technique used in many pharmaceutical applications including potency/purity/performance assays, pharmacokinetics/ bioanalytical testing, purification, high-throughput screening (HTS), in process control (IPC) monitoring and quality control (QC) testing [1-6]. The pharmaceutical industry is the major consumer segment of HPLC [7] and has been the primary driving force for higher throughput and performance. This article provides a brief review of significant developments in HPLC impacting pharmaceutical analysis in the last decade. Topics discussed are listed below and selected items are shown in a representative timeline in Figure 1.

- Instrumentation: Ultra-high pressure LC (UHPLC) going mainstream.
- Columns: Sub-2 μm, sub-3μm core-shell and hybrid particles; novel bonding chemistries; hydrophilic interaction chromatography (HILIC); immobilized polysaccharide chiral phases; columns for biomolecules and biopharmaceuticals.



pharmaceutical analysis in the last decade.

Others: Liquid chromatography - mass spectrometry (LC/ MS) - particularly high-resolution MS (HRMS) or hybrid MS; charged aerosol detector (CAD); automated method development systems (AMDS).

Our goal is to provide the busy pharmaceutical scientist with a concise yet comprehensive update of important HPLC developments, with each topic supported by brief descriptions of practical benefits/applications, critical commentaries from a user's perspective, and key references.

## Ultra-High Pressure Liquid Chromatography (UHPLC)

The "revolution" in ultra-high pressure LC (UHPLC) began in 1997 with the proof-of-concept study by Professor James Jorgenson [8], followed by the first commercial system introduced in 2004 [1, 9-11]. Today, the transformation from HPLC to UHPLC is mostly complete with all major manufacturers having some type of UHPLC offerings. Detailed reviews of UHPLC systems, columns and applications are available elsewhere [4, 10-11]. Fundamentals, benefits, potential issues and best practices of UHPLC in pharmaceutical analysis are well documented [4, 11-14].

Table 1. Prominent Characteristics and benefits of UHPLC	
System characteristics of UHPLC	Range and Comment
High pressure limit	15,000 to 19,000 psi (1000 to 1250 bar) with flow rate limits of 2 to 5 mL/min. Compatible with conventional and sub-2 µm particle columns.
Low system dispersion	Instrumental bandwidth of 5 to 20 $\mu$ L (4 $\sigma$ ) depending on configuration. System band broadening reduced by using smaller connection tubing <0.005" I.D. and small UV flow cells (0.5 to 2 $\mu$ L). Compatible with columns down to 2 to 3 mm ID.
Low gradient dwell volume	100 to 400 μL (higher for quaternary pumps). Compatible to high throughput screening (HTS). Small dwell (mixing) volumes may negatively impact UV detector noise.
Others	Fast injection cycle (~20s) and detector response, and high acquisition rate (>40 pt/s) for HTS. Compatibility to existing HPLC methods desirable (e.g., flow range, column oven size, sample loop).
Benefits of UHPLC	Comment
High throughput	Increase throughput by 3 -10 fold vs. conventional HPLC while maintaining similar resolution, e.g., 5 min vs. 20 min purity analysis.
Rapid method development	Fast analysis with short columns is ideal for rapid column and mobile phase screening and method optimization.
High resolution	Increase resolution by up to 3 fold vs. HPLC, e.g., Peak capacities (P <sub>c</sub> ) 400 – 600 vs. ~200 for HPLC.
Solvent saving	Typical 5 to 15 fold reduction vs. HPLC due to shorter analysis time and use of smaller ID columns.
Higher sensitivity	3 to 10 fold increase of mass sensitivity (reduction of sample amounts injected). Use of long-pathlength UV flow cells (50 – 60 mm) can increase concentration sensitivity up to six times
Higher precision	Significant increase of retention time (2 - 3 fold) and peak area precision (<0.1% RSD achievable at injection volumes >1 µL)

Table 1 summarizes the prominent characteristics and benefits of UHPLC. Note that higher system pressures allow the use of columns packed with smaller particles (e.g., sub-2 µm) for faster analyses (Figure 2) or superior separations of complex samples (Figure 3). All UHPLC have low system dispersion ( $\sim$ 10 to 20  $\mu$ L for 4 $\sigma$  bandwidths) from the use of improved injectors, smaller ID tubing (<0.005") and smaller UV detector flow cells (0.5 – 2  $\mu$ L) [1, 9-11]. The use of 2.1 - 3.0 mm ID columns packed with sub-2 µm or sub-3 µm particles is typical. Other important system characteristics include smaller system dwell volumes (0.1 – 0.3 mL) and faster detector response/data acquisition (>40 pt/s) for high-throughput applications.

The benefit of faster analysis of UHPLC vs. conventional HPLC is illustrated in Figure 2 in a method transfer case study for a drug product. Using a geometrical scaling of column and operating parameters from HPLC to UHPLC [11, 14, 15], a reduction of analysis time up to tenfold with similar resolution, is not unusual. This benefit of "faster analysis with good resolution" provides the primary incentive for most users to consider the purchase of the more expensive UHPLC equipment.

Another important benefit of UHPLC is its superior separation of complex samples. This aspect is often overlooked and appears to be "under-reported" in the literature [17-18]. Peak capacities (P) in the range of 400 to 1000 in a reasonable time span (~1h) have been demonstrated using UHPLC. Peak capacity is the number of peaks that can be resolved in the chromatogram with a resolution of 1.0; typically ~200 for conventional HPLC [1]. For the first time ever, UHPLC can offer more effective assays, in a single dimension, for complex pharmaceuticals, natural materials, and other difficult sample matrices [4, 11, 17-18]. Figure 3a and 3b show two applications (a plant extract and a protein tryptic digest) to illustrate the current capability of UHPLC in high-resolution separations with Pc >300.

Other benefits of UHPLC include substantial solvent savings (5-15 fold), increased mass sensitivity (3 - 10 fold) and precision performance for

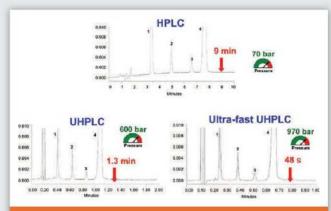
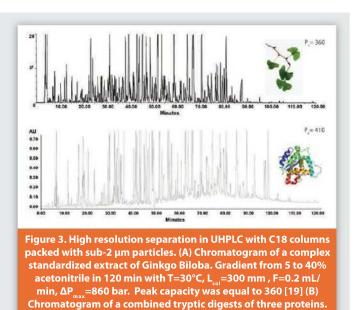


Figure 2. An example of a method transfer for a quality control assay of a commercial pharmaceutical formulation (Rapidocaïn) using geometrical scaling from conventional HPLC to UHPLC. Peak identification: 1: methylparaben, 2: 2,6-dimethylaniline, 3: propylparaben, 4: lidocaïne. HPLC conditions: column RP18 150x4.6 mm, 5  $\mu$ m, F=1 mL/min,  $V_{_{\rm inj}}$ =20  $\mu$ L. UHPLC conditions: column RP18 50x2.1 mm, 1.7 μm, F=600 μL/min, V<sub>...</sub>=1.4 μL. Ultrafast UHPLC conditions: column RP18 50x2.1 mm, 1.7 μm, F=1000  $\mu$ L/min,  $V_{ini}$ =1.4  $\mu$ L [16].



Gradient from 5 to 60% acetonitrile in 120 min with T=30°C,  $L_{col}$ =40 cm, F=0.16 mL/min,  $\Delta P_{max}$ =940 bar. Peak capacity was equal to 410 [20].

both retention times (2 - 3 fold) and peak areas (<0.1% RSD) [11, 14, 18]. Note that reports on UV detection sensitivity increase with UHPLC are often misleading because mass sensitivity (amount of analytes injected) is primarily related to column void volumes. UHPLC typically does not increase concentration sensitivity (the most desirable kind of sensitivity) because one does not expect that the use of small flow cells can actually increase signal/noise ratio unless extended pathlength flow cells (e.g., 60 mm) are used. Potential issues such as viscous heating, baseline perturbation from pump blending, and method transfer have been described and can be instrument-specific [4, 11, 18]. In general, these technical issues are well understood and can be readily mitigated by judicious choice of system configuration (e.g., mixer volume). Nevertheless, the implementation of UHPLC in QC labs remains to be time-consuming due to training, compatibility to "validated" data systems, and other method transfer issues [4, 18].

One often hears arguments that UHPLC is perhaps not needed because other approaches (high-temperature LC, core-shell columns or 2-D LC) can be more cost-effective for enhancing speed or resolution. The reasoning is not truly valid since UHPLC can be used in combination with one or more of these approaches with superior results than those from conventional HPLC, as they are options rather than alternatives. Also, the term "UHPLC" may eventually go away in a few years since all newer HPLC will be UHPLC.

# HPLC Column and Stationary Phase Developments

HPLC column is the heart of the chromatographic system. The pharmaceutical industry has been the primary driver for HPLC columns towards higher speed, resolution, and better peak shapes for

Table 2. Significant HPLC column developments impacting pharmaceutical analysis	
Significant HPLC column developments	Commentary
Sub-2 μm particles	Fully porous sub-2 µm particles yield lower plate heights and higher efficiency. >80 chemistries available including those for ion-exchange and size exclusion chromatography from 10+ providers.
Sub-3 μm core-shell particles	Superficially porous sub-3 µm particles offer improved mass transfer and lower plate heights operating at lower column pressures. Increased usage in pharmaceutical analysis.
Hybrid silica particles	These innovative particles provide enhanced chemical stability to the column (extended pH range, temperature and pressure capability; lower silanophilic activity).
Novel bonding chemistries	Polyfunctional silane chemistry; polar embedded bonded phases, pentafluorophenyl (PFP), phenyl-hexyl and charged surface hybrid (CSH) phases offer "orthogonal" separations and enhanced selectivity to traditional C18 bonded phases.
Hydrophilic interaction chromatography (HILIC)	Alternative "orthogonal" separations to reversed phase LC useful for highly polar compounds and secondary metabolites.
Immobilized polysaccharide phases for chiral separations	The polysaccharide chiral stationary phases are immobilized to make these popular columns more versatile and robust.
Columns for biomolecules	Innovative core-shell and UHPLC wide-pore phases are available for improved analysis of proteins and therapeutic biomolecules (e.g., mAbs).

basic analytes. In addition, QC laboratories have demanded improved column batch-to-batch reproducibility. From the 1970s to 1990s, there were steady quality improvements of the packing materials accompanied by a gradual reduction of "standard" particle sizes from 10 to 3  $\mu m$  [1-3]. The introduction of high-purity type B silica materials (with low metallic content) in the late 1980s was a huge step and resulted in reduced silanol activity and substantial improvements in lot-to-lot consistency [21]. The use of high-purity silica is now the norm for all modern silica-based columns.

Recent developments in HPLC column technologies have been extensively reviewed [22]. Table 2 summarizes those with significant adoption rates by the industry for increasing productivity, stability, selectivity or retention (sub-2 µm, core-shell particles, hybrids, novel bonding chemistries, HILIC) or specialized applications (chiral compounds or biomolecules).

#### Sub-2 µm Particles

The use of very small particles for fast and efficient separations was predicted in 1956 by Van Deemter [23]. Typical particle size of packings has been decreasing in the past five decades cumulating to the use of sub-2 µm microparticulate silica in the early 2000s. As predicted, these particles (e.g., 1.7 µm) yield excellent efficiency performance (~280,000 plates/m or plate height of ~4 μm). However, columns packed with sub-2 µm particles generate high back pressures and are typically packed in a 2.1-mm ID format to reduce efficiency loss from viscous heating effects [4, 9, 14]. The system requirements for high pressures and low dispersion (to reduce extracolumn band broadening) were responsible for the current characteristics of modern UHPLC systems. Further reduction of particle size to less than 1.5 µm may be advantageous for even higher speed and performance. However, it must be accompanied by a substantial increase of system pressures and a reduction of column ID to capillary formats.

#### Core-shell Particles

The concept of fused-core or core-shell particles for reducing resistance to mass transfer was first described by Kirkland [24]. The first core-shell particles had these characteristics: 2.7 µm superficially porous silica materials with nonporous cores (1.7 μm) and porous shells (0.5 µm thick) [24-25]. These sub-3 µm particles appear to have similar efficiencies as the sub-2 µm fully porous materials but generate much lower pressure drops. The exceptional performance may be due to their shorter diffusion paths of the shells and/or the narrower distribution of the packings. Core-shell columns are rapidly gaining wide acceptance for fast separations (HTS, IPC and cleaning verification) and for biomolecules [25-27]. Myriad bonded phases and particle sizes (1.3, 1.7, 2.6, 2.7 and 5 µm) are currently available from an increasing number of manufacturers (6+). We fully expect these columns to be highly competitive in all applications to those from porous microparticulates [25].

## Hybrids

The concepts of hybrid particles with organic groups incorporated into the inorganic silica matrices were first described by Unger [28] in late 1970s, though the first commercial columns with methyl groups only became available in 1999 [29]. Bonded phases from these hybrids were demonstrated to have superior pH stability (from 1 to 12 with novel bonding chemistries vs. 2 to 8 for conventional silica with traditional monofunctional bonding), and lower silanophilic activity. A second-generation bridged ethylene hybrids (BEH) was introduced in 2005 and enjoyed immediate "mainstream" success, particularly for applications with high-pH mobile phase applications and UHPLC [9-10, 30].

#### Novel Bonding Chemistries

While the traditional monofunctional C18 silica-based bonded phase remains to be the main "staple" with perhaps the best batch-to-batch reproducibility [1, 3], newer bonding chemistries have brought benefits such as a wider pH stability range (from polyfunctional silane chemistry or silanes with isopropyl protective groups) and enhanced selectivity for difficult separations (polarembedded, phenyl, hexylphenyl, cyano, pentylfluorophenyl bonding) [1,3]. One recent innovation, termed charged surface hybrids (CSH) technology introduced in 2010, appears to have gained immediate acceptance for pharmaceutical analysis due to improved peak shapes of highly basic analytes under acidic, lowionic-strength mobile phases conditions (e.g., 0.1% formic acid) [31]. The proprietary technology involves the introduction of a low level of positive charges to the surface of the stationary phases. This is somewhat analogous to the addition of an amine additive such as triethylamine to the mobile phases in earlier days which is no longer acceptable in LC/MS due to ion suppression. Figure 4 illustrates the benefits of CSH particles in peak shape performance vs. comparable contemporary bonded phases [31]. Note that acidic compounds will have comparatively higher retention and increased peak tailing on CSH columns.

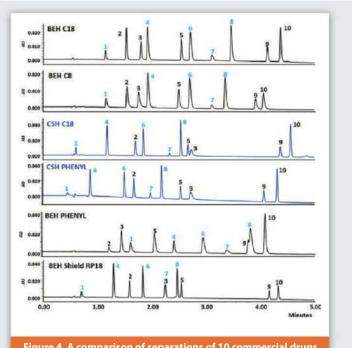


Figure 4. A comparison of separations of 10 commercial drugs with acidic, basic and neutral functional groups on BEH and CSH analytical columns using gradient elution with mobile phase containing acetonitrile and 0.1% formic acid. Peaks 1, 4, 6 and 8 were basic drugs. Reprinted from [31], with permission.

## *Hydrophilic Interaction Chromatography (HILIC)*

The retention of many highly polar compounds is simply unachievable or problematic due to phase collapse (bonded phase "dewetting") under reversed-phase LC (RPLC) conditions in mobile phases with low organic contents [1-3]. The HILIC mode, first developed by Alpert in 1990s [32], uses a hydrophilic stationary phase (silica, diol, cyano, amide, ztwitterionic) with an RPLClike aqueous buffer and acetonitrile mobile phases, has enjoyed increasing popularity for the analysis of polar drugs, secondary drug metabolites, amino acids, peptides, neurotransmitters, oligosaccharides, carbohydrates, nucleotides, or nucleosides [33]. The actual retention mechanism in HILIC can be the "partitioning" of analyte molecules to the water layer adhering to the hydrophilic bonded groups. Other prominent benefits of HILIC include "orthogonal" selectivity to RPLC where sample preparations are compatible to both modes, higher electrospray ionization sensitivity for MS (5-15 fold), and lower operating pressures vs. RPLC.

## Immobilized Polysaccharide Chiral Stationary Phases

Improved versions of the highly successful coated polysaccharides chiral stationary phases (CSPs) were available in the late 2000s. They brought similar versatility as the earlier coated CSPs but are more robust to aggressive solvents and can be used in normal phase, polarorganic and reversed-phase modes [5, 34].

## Columns for Biomolecules

Wide-pore silica and polymeric packings first available in 1980s were effective for separations of large biomolecules [1, 35]. With the advent of recombinant proteins as biopharmaceuticals such as monoclonal antibodies (mAb), the need for detailed characterization by HPLC or capillary electrophoresis for QC purposes has become more pressing [36]. Recent developments of sub-2 µm microparticulates and coreshell wide-pore particles as well as several innovative ion-exchange and size exclusion phases have proved useful for separations of these large biologics [37].

# High Resolution Mass spectrometry (HRMS), Charged Aerosol Detection (CAD) and Automated Method Development Systems (AMDS)

Other high-impact developments in the past decade were in the HPLC detection and automation areas.

## *High Resolution Mass Spectrometry (HRMS)*

The combination of HPLC with mass spectrometry (LC/MS) has been touted as the perfect analytical tool, combining the separation power of HPLC and the unsurpassed sensitivity/selectivity of MS. LC/MS is the preferred technique for identification of impurities and degradants, HTS in drug discovery, bioanalytical assays (LC/MS/MS for drugs and metabolites in biofluids), and in-progress monitoring during process scale-up for the synthesis of drug substances [1-2, 5, 38]. LC/MS is becoming the standard platform technology for cleaning verification of highly-potent drugs [26] and determination of potential genotoxic impurities [2]. The last decade has seen the rapid evolution of HRMS (such as time-of-flight (TOF), OrbiTrap MS) and hybrid MS (such as Quadrupole-TOF or ion trap-OrbiTrap). The combination of HRMS with UHPLC and 2-D LC has enabled active research in metabonomics, proteomics, de nova protein sequencing, and characterization of biopharmaceuticals [5, 38]. Perhaps the most exciting opportunity for LC/MS lies ahead as a generic platform technology for the determination of disease biomarkers and clinical diagnostics in the emergent field of personalized medicine.

## Charged Aerosol Detection (CAD)

The lack of an ideal universal detector is often cited as a limitation of HPLC, though the UV/Vis detector comes fairly close for chromophoric compounds. The refractive index detector is not gradient compatible and does not have sufficient sensitivity [1-3]. Evaporative light scattering detection (ELSD) using nebulizer technology with laser light scattering detection is an option and is gradient compatible but has been recently surpassed by CAD (uses nebulizer with corona discharge detection), which offers better sensitivity (low ng) and improved linearity. CAD is becoming a mainstream detector for HTS in medicinal chemistry, reaction monitoring, and raw material/excipient testing [39].

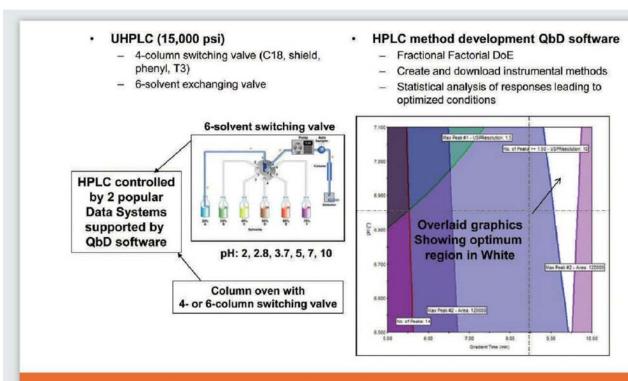


Figure 5. Schematic diagram of an example instrument configuration of an automated method development system using QbD software and a resulting graphics showing a design space with an operating region for optimum separations (white color).



Our goal is to provide the busy pharmaceutical scientist with a concise yet comprehensive update of important HPLC developments.

#### Automated HPLC Method System (AMDS)

HPLC method development for complex mixtures is time-consuming due to the need to optimize many operating parameters (column dimension; type of bonded phase and mobile phase A and B (organic solvent/buffer type, pH, and ionic strength), gradient time and range, column temperature, and flow rate) [1, 5, 6]. A common example is the stability-indicating or purity assay of active pharmaceutical ingredients (API) in which all impurities and degradants must be separated for accurate quantitation by UV detection. Software or automated systems based on simulation, prediction, simplex optimization, and column/mobile phase screening have been available to facilitate HPLC method development for many years. Continued improvements have enhanced their capability and ease-of-use though they never appeared to be very popular [2, 6]. The latest market entry was an add-on software package compatible to two of the commonly used chromatography data systems. The software addresses the most time-consuming portion of the HPLC method development process (optimization) by automating method sequence from a user-defined design space using principles of Design of Experiments (DoE) and Quality by Design (QbD) [40-41]. This software can also perform statistical analysis and display optimum conditions after importing of the completed sequence results. Figure 5 illustrates the concepts and salient features of this software as implemented in our laboratory. Since concepts of DoE and QbD are well accepted, initial customer interest appeared to be high. AMDS is particularly useful to laboratories specializing in method development to support earlyphase drug development [2, 6]. For instance, to support process developments for API synthesis and drug product manufacturing from complex drug molecules with multiple chiral centers, as many as 10-20 HPLC methods (achiral and chiral methods for raw materials, starting materials, intermediates, final drug substances and drug products) are typically developed in quick succession.

## Summary and Conclusions

In summary, HPLC remains a highly dynamic field with numerous innovations in instruments, column technologies, and approaches in recent years. Pharmaceutical scientists are early adopters and  $beneficiaries\ of\ these\ newer\ technologies\ for\ research, development$ and quality control. UHPLC is becoming the standard HPLC platform with rapid adoption by research & development, albeit slower implementation in QC labs. Newer column technologies allow faster and more efficient analysis of complex samples, chiral molecules and biomolecules. Finally, the rapid advancements of UHPLC and 2-D LC in combination with high-resolution MS have revolutionized life science research and promise to have even greater impact for clinical diagnostics. These developments are welcomed progress for the analytical chemist working in this rapidevolving world of drug development.

As a perspective for LC in the next years, some innovative works can be mentioned [42, 43]. As described by Professor Jim Jorgenson [42]: "Moving to still higher pressures (50,000 psi) will enable the use of smaller particles and/or longer columns, and yield faster and better separations. This will almost certainly require the use of sub-mm bore (capillary) columns due to issues with heat generation and dissipation. This won't be easy, but the separation potential in terms of high speed with high resolution is enticing." Alternatively, the use of extremely uniform packing of sub-micron silica particles in capillaries could be employed to generate plate heights well below 1 µm and impressive separations of protein variants [43]. These two research findings are unlikely to be ready for mainstream analytical work or QC very soon but the first results are encouraging and thus worth mentioning here.

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